

Green Synthesis of Silver Nanoparticles using Neem and Collagen of Fish Scales as a Reducing and Stabilizer Agents

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Abstract

This study was conducted to synthesize and characterize silver nanoparticles (AgNPs) using a rapid green synthesis approach. Antibacterial properties of AgNPs were evaluated. Extract of *Melia Dubia* (Neem) and collagen produced from the fish scales were used as reducing and stabilizer agents respectively. Uv-Vis Spectroscopy, Scanning Electron Microscopy (SEM), Energy-Dispersive X-Ray Spectroscopy (EDX), Fourier-Transform Infrared Spectroscopy (FTIR), and Dynamic Light Scattering (DLS) were used to characterize the synthesized AgNPs. Evaluation of the synthesized AgNP's antibacterial activity was conducted, with *Staphylococcus Aureus* (*S. aureus*) as Gram-positive and *Escherichia Coli* (*E. coli*) as the negative bacteria. The peak of absorbance for the synthesized AgNPs was at 454 nm, indicating conformed AgNPs. SEM image showed semi-evenly distributed rod shapes. The EDX data revealed presence of the metallic silver. Meanwhile, FTIR analysis indicated presence of C₂H₂, C=O, N-H groups. DLS showed an average size of 437.6 nm. XRD showed calculation for particle size using $d = K\lambda/\beta\cos\theta$. Average size of AgNPs was 141.81 ± 5 nm. AgNPs also displayed tangible antibacterial activity towards the *S. aureus* and pathogenic *E. coli*. AgNPs were successfully synthesized and evaluated for their antibacterial properties. The outcomes being the multifaceted biological activities alongside application of biocompatible green nanoparticles, is discoverable in the field of Nanomedicine.

Keywords: Silver nanoparticles, Fish Scales Collagen (FsCol), *Melia Dubia* (neem), Antibacterial

1. Introduction

AgNPs are common nanoparticles, vastly used in cleaning agents, commercial industry, food storage, healthcare products, packing and textile coating. Their antimicrobial property against bacteria, fungi, and viruses, makes them suitable for several environmental applications besides being used extensively in products specific to biomedical purposes such as topical creams, antiseptic sprays, and wound dressings. This is due to their capability in disabling a microorganism's enzymatic activities by disrupting the membrane of the pathogenic organisms (Haghighi et al., 2017; Pugazhendhi et al., 2018).

Many techniques were introduced to synthesize AgNPs, such as thermal decomposition, chemical reduction of silver ions (Ag⁺), pyrolysis microwave irradiation, electrochemical methods, and laser ablations (Beyene et al., 2017; Sundarrajan et al., 2019). Among all, the most traditional way to synthesize AgNPs is via chemical reduction technique. The agents commonly used lessen the chemical are sodium borohydride and sodium citrate (Ojo et al., 2017). Traditional chemical synthesis stipulates the necessity to have metal precursor, reducing agents and capping/stabilizing agents to prevent vigorous spreading of

the nanoparticles (Chowdhury et al., 2015). The traditional ways are claimed to be more effective in controlling distribution and reproducibility of the nanoparticle size as they require greater input of energy alongside presence of toxic chemicals. Under certain circumstances, it may require occurrence of controlled pressure and temperature. This may lead to higher cost and environmental contamination.

"Green chemistry" was introduced as an economically feasible option and is known to be a biogenic way to synthesize metallic nanoparticles and AgNPs (Ahmed et al., 2019). This approach is known to be a biogenic synthesis of the metallic nanoparticles. This approach is reputable for its simplicity, environmental-friendliness, cost-effectiveness and scale-up efficiency. The procedure is carried out at room temperature and pressure, within aqueous ambience, without organic solvents (Sportelli et al., 2018). It is also conducted via biological systems, which includes algae, bacteria, plants, fungi, diatoms, and yeast. The simplicity of the biogenic way made it possible for plant extracts to be used to eradicate the procedure of sustaining cell cultures. Metallic nanoparticles that are based on plant-mediated synthesis are also less biohazardous (Mudhafar et al., 2020).

Recently, plant extracts have gained much interest. They are commonly used as agents to reduce and cap

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metallic nanoparticles mainly due to their advantageous potentials which include fast, simple, economically feasible green technique, and environmentally-friendly nature (Ajitha et al., 2016). The phytochemicals of the plants play the role of tough reducing agent, resulting in development of capped nanoparticles. This signifies that plant extract is a one-step natural capping and reducing agent, reducing not just the cost but the use of chemical reagents too (Govindarajan et al., 2017). In addition, chemical compounds are replaced with phytochemicals' surroundings, while organic solvents are replaced by aqueous surroundings (Jyoti et al., 2016). Today, distinct parts of plants, like its fruits, bran, bark, flowers, and leaf, are used in metallic nanoparticle synthesis (Shaik et al., 2018) as they possess medicinal value attributed to presence of amino acids, phenolic compounds, such as polyphenols, flavonoids, terpenoids, and vitamins (Mudhafar et al., 2019).

Neem is a part of the *Melia* genes belonging to the *Meliaceae* family. It is greatly distributed in India and Malaysia. The Indian neem have been extracted and used to produce nano-silver, showing good result when AgNO_3 was used for anticancer activity. Kathiravan et al., (2014), have synthesized the AgNPs for anticancer. However, there has yet to be any report of AgNPs synthesis from neem leaves and its antimicrobial activity from neem extract.

Collagen is a known biocompatible polymer that is water-soluble, and synthetic, evidenced by its broad use for biomedical purposes. Among such purpose includes the use of nanoparticles' coating to enhance spreading of nanoparticles in an aqueous medium and their penetration into the mucus layer. Furthermore, collagen can guard nanoparticle against immune system clearance by avoiding the opsonization of nanoparticle in the blood, lessening the toxicity of nanoparticles, and increasing the lifetime of the nanomaterial within the circulation system (Sawadkar et al., 2019; Mudhafar et al., 2020). Additionally, collagen can also be used as a stabilizing agent (Tarannum et al., 2015). For the purpose of this study, Fish Collagen was used as safe stabilizing agent for AgNPs.

The AgNPs' biogenic synthesis was studied with the one-pot green protocol because of its advantageous traits. Meanwhile, the reducing agent used was from neem extracts. To enhance the capability of the green neem that synthesized the AgNPs, the nanoparticles were stabilized with collagen. Different analytical methods were conducted to characterize the AgNPs. The classification of the AgNPs' infusion of antimicrobial activity towards a few Gram-positive and Gram-negative pathogenic bacterial strain was conducted based on *in vitro* evaluation. Outcome from a simple green protocol revealed that the commercial green neem extract led to successful development of plant-mediated AgNPs. Furthermore, the nanoparticles' antibacterial activity against varying strains of bacteria was shown at concentration. These support outcomes from the previous study, where biosynthesis of nanometals had strong antibacterial effects.

2. Materials and methods

2.1. Materials

AgNO_3 was obtained from Bendosen Company. Neem leaves were sourced locally from Perak, Malaysia. Collagen was extracted from fish scales and were produced in the Chemistry Department laboratory, UPSI, Perak, Malaysia. The broth and nutrient agar were bought from Merck.

2.2. Methods

2.2.1. Neem leaf preparation

Fresh neem leaves were collected from Perak. The leaves were cleaned several times to eradicate dust and fungi after which it was sun-dried for one week to completely remove moisture. The dried neem leaves were then crushed and converted to powder form. A total of 10g powder was extracted with 100 ml distilled water in a 250 ml conical flask. Subsequently, the powder was heated for 10 minutes, cooled and filtered to obtain a crude extract before being incubated at 4 °C.

2.2.2. Green synthesis of AgNPs

1 mM of AgNO_3 was dissolved in 90 ml distilled water and the solution formed was transferred into a 250ml conical flask. 10 ml of crude extract was then added into the conical flask and stirred thoroughly. Then, 0.1g of collagen was added into the conical flask and the mixture was again stirred thoroughly. UV-Vis Spectroscopy was then utilized to analyses development of AgNPs. The solution containing AgNPs was centrifuged for 10 minutes at a rate of 14000 rpm. Meanwhile, the supernatants were discarded, and the NP portion was sterilized in distilled water. This aforementioned step was repeated three times to eradicate unused biomaterial and impurity. The freeze-drying method was used to obtain dry particles for the purified sample of dry particles (Joseph et al., 2015).

2.2.3. Characterization of AgNPs

Absorbance between 200nm and 800 nm was conducted using the UV-Vis Spectroscopy Agilent Cary 60 UV Spectrophotometer. EDX, SEM, and DLS were performed respectively via Bruker X Flash 6110, Nova Nanosem 450.

2.2.4. Antibacterial activity

Paper disk diffusion test was employed to investigate antibacterial activities of *E. coli* and *S. aureus*. Nutrient agar media was used to cultivate bacteria. 10 µg of AgNPs was saturated in 1 ml of distilled water and tested against the bacteria. Neem were also examined. Next, a total of 10, 20, 40, and 80 µM/ml of AgNPs were dissolved in distilled water and poured onto a 6 mm disk filter [24]. Upon 24 hours incubation, inhibited zones were measured and the magnitude of antibacterial impacts on *S. aureus* and the *E. coli* were determined.

3. Results and Discussion

3.1. UV-Vis Spectroscopy

UV-Vis Spectroscopy was used to corroborate stability and development of AgNPs in a solution. Figure 1 displays the extinction spectra of AgNPs. As observed, the

absorption band of synthesized AgNPs was 454 nm. Many studies have documented absorbance of AgNPs, whereby peak of surface plasmon for AgNPs sized <100 nm was located between 400–450 nm (Rathod et al., 2016).

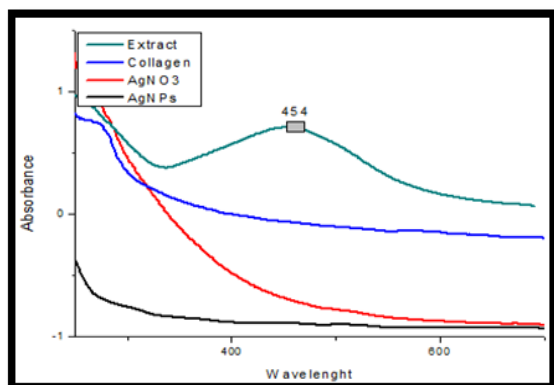


Figure 1. UV-Vis Spectroscopy of AgNPs, AgNO₃ and FsCol

3.2. SEM

SEM was utilised to classify the parts of synthesized AgNPs. SEM's image displays the AgNPs' constant spreading. It was discovered that the AgNPs had a rod-like shape with smooth morphology, almost the same shape as the AgNPs when using a microwave. Figure 2 displays SEM image such as that of the prepared powdered AgNPs. The magnification specification for SEM image was 100000x for 500 nm image size, which is a considerably big for AgNPs. Especially since the formation of AgNPs were considerably shorter compared to that mentioned in literature; that the size of the Silver Nanoparticles will decrease with increased formation time (Dehnavi et al., 2013).

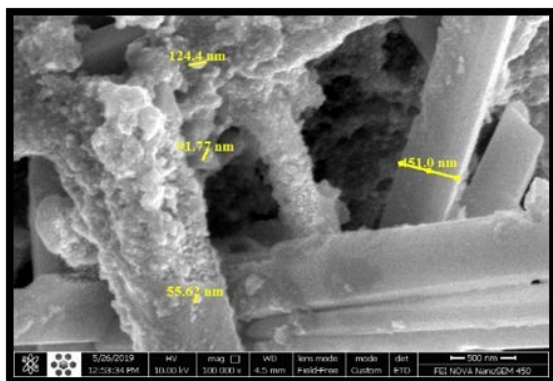


Figure 2. SEM for the morphology and measured of AgNPs

3.3. FTIR

FTIR was used to evaluate factional groups of fish collagen and neem leaves in comparison with factional groups of synthesized AgNPs. This is to gain insight on factional groups surrounding the silver nanoparticles. As shown in Figure 4, the FsCol reveals two major peaks, which are 3441 cm⁻¹ for the NH (amide A) and 1646 cm⁻¹ for CO stretching (amide I) (Hamidi et al., 2019). Peaks of the four miners were monitored at 1548 cm⁻¹, 1466 cm⁻¹, 1392 cm⁻¹, and 1242 cm⁻¹. 1548 cm⁻¹-1466 cm⁻¹ refers to the N-H bend coupled with C-N stretch (Amide II), and 1392 -1242 cm⁻¹ refers to the N-H bending (Amide III) [23]. Neem leaves were characterized with two major

peaks located at 3296 cm⁻¹ and 1629 cm⁻¹. The peak at 3296 cm⁻¹ refers to the stretch of N-H and O-H due to the presence of water, alcohols, and phenols (polyphenols) in neem extract (Philip, 2010). Amide I was observed at 1629 cm⁻¹, while minor peaks were observed at 2115 cm⁻¹, 618 cm⁻¹, 480 cm⁻¹, and 333 cm⁻¹. At 2115 cm⁻¹, the bands appeared parallel to the present Alkyne group (C₂H₂) because of neem extract phytoconstituents (Mudhafar et al., 2019). At 618 cm⁻¹, the band matched the C-H bonds in the phenolic rings while the band at 480 cm⁻¹ were attributed to the existence of alkyl halides.

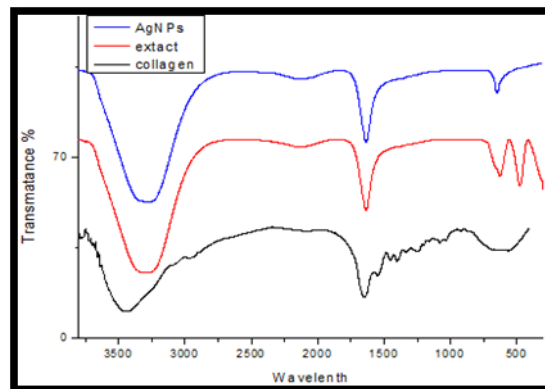


Figure 3. FTIR spectrum of FsCol, *M. Dubia*, and AgNPs

Among AgNPs, two major peaks were observed in the FTIR graph; 3315 cm⁻¹ and 1641 cm⁻¹. At 3315 cm⁻¹, the band is referring to the stretch of O-H and N-H (amide A), while at 1641 cm⁻¹, the band refers to the similar stretching of CO that belonging to amide I. Both these bands were observed in the fish collagen and neem extract. The band located at 2128 cm⁻¹, refers to alkyne group (C₂H₂) from the neem extract phytoconstituents. At 1361 cm⁻¹, the band corresponds to N-H bending (Amide III) from fish collagen. Two peaks at 608 cm⁻¹ and 689 cm⁻¹ indicates C-H bonds in phenolic rings. Table 1 shows all details for AgNPs, plant extract and FsCol (Singh et al., 2016; Mudhafar et al., 2021).

Table 1: Factional groups in AgNPs in FTIR

Factional groups	Wavenumber cm ⁻¹		
	FsCOL	Plant extract	AgNPs
N-H (stretch)	3446 cm ⁻¹	3296 cm ⁻¹	3301 cm ⁻¹
Alkyne group (C ₂ H ₂)	--	2115 cm ⁻¹	2140 cm ⁻¹
C=O Stretching (amide I)	1646 cm ⁻¹	1629 cm ⁻¹	1668 cm ⁻¹
N-H bends couple C-N stretch (Amide II) CN	1531 cm ⁻¹	--	--
Amide III N-H bend	1348 cm ⁻¹	--	1366 cm ⁻¹
Amide III N-H bend	1242 cm ⁻¹	--	1227 cm ⁻¹
O-H	--	618 cm ⁻¹	634 cm ⁻¹
Alkyl halides	--	480 cm ⁻¹	488 cm ⁻¹

3.4. DLS

Zeta sizer shows the peak of AgNPs. Based on the pattern, the AgNPs synthesized through this technique had shown a 0.249 polydispersity index (PDI) with a Zeta average diameter of 437.6 nm. The size of the zeta sizer measures is slightly bigger than the size of the particle as measured in SEM micrographs due to DLS method which measures the hydrodynamic radius (Ahmed et al., 2017). In addition, DLS also measures attached surface proteins,

carbohydrates, and other cellular materials. However, they could not be kept in a vacuum under an electron beam as it can also be associated with the consistency of biosynthesis (Liu et al., 2019). The DLS graph of synthesized AgNPs showed two peaks, a strong peak at 480.7 nm with 98% intensity area and a weak peak at 5256 nm with 2% intensity area. Meanwhile, the Zeta average diameter was 437.6 nm, as shown in Figure 5.

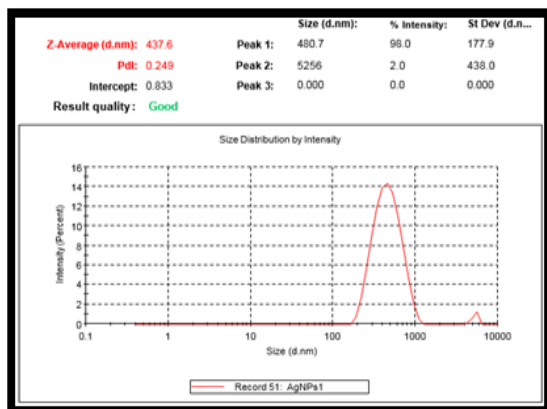


Figure 4. DLS analyses of AgNPs

3.5. Antibacterial activities

The AgNPs' antibacterial properties were recorded according to disk diffusion method. A ruler was used to measure the AgNPs' ability to prevent bacterial growth against both bacteria (Anandalakshmi et al., 2016). Distilled water was used for negative control, while ampicillin was used and tested against bacteria for positive control. Figure 5 shows the inhibition zones of *E. coli* and *S. aureus* bacteria. The distilled water inhibition appeared zero in both bacteria. The inhibition zones of the extract were 12.2 and 10.4 mm, in contrast to *E. coli* and *S. aureus*, respectively. The inhibition zones of ampicillin were 28.6 for *E. coli* and 22.1 for *S. aureus*. Many AgNPs' concentrations resulted in good inhibition zones against both bacteria. The inhibition zone increased with AgNPs concentration. Table 2 shows details of AgNPs inhibition and control.

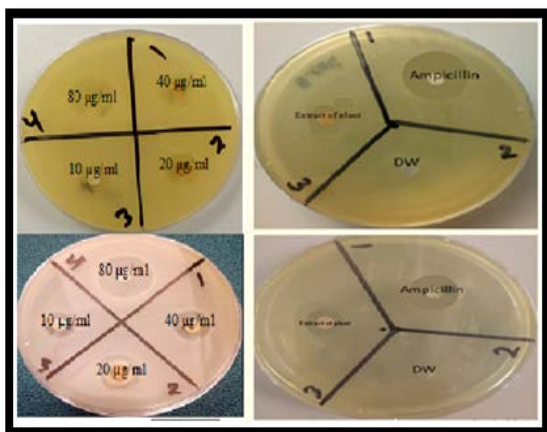


Figure 5. Antibacterial activities of A) AgNPs and control samples against *E. Coli* and B) AgNPs against *S. Aureus*

4. Conclusions

The characteristics of biosynthesis methods, such as safety, low cost, and environmental friendliness are deemed attractive by researchers. Green synthesis method was employed in this study to synthesize AgNPs from biological sources. Neem leaves were utilized as natural source. The use of UV-vis, FTIR, SEM, EDX, and DLS resulted in successful synthesis and categorization of AgNPs. The morphology and particle size were also investigated. These techniques revealed a rod shape with average sizes ranging from 150 nm to 430 nm. The antimicrobial characteristics of the synthesized AgNPs were also examined using the Gram-positive and Gram-negative bacteria. To which it was discovered that AgNPs was highly efficient as an agent for microbes and has great potential for use as antibacterial agent within the medical field.

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