

# An Assessment of Proximate Composition, Antioxidant Activities and LC/MS Based Phytochemical Profiling of Some Lichen Species Collected From Western Ghats of Southern Part of India

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## Abstract

The lichen samples of *Parmotrema tinctorum*, *Pseudocyphellaria aurata*, *Ramalina taitensis*, *Usnea bismolliuscula* are being used as nutraceutical agents in indigenous system of medicine. The main objectives of this study were assessing the proximate composition in lichen samples of the extracts and their antioxidant properties. Using Liquid Chromatography Mass Spectrophotometric analysis with chemoprofiling of the studied lichen compositions were also done. The petroleum ether, acetone, methanol and water extracts of test samples obtained from Soxhlet apparatus were assessed for *in vitro* antioxidant activities using 1,1-diphenyl-2-picrylhydrazine, Superoxide anion radical, 2, 2'-Azino-Bis-3-Ethylbenzothiazoline-6-Sulfonic Acid scavenging and Phosphomolybdenum reduction assays. The methanol extract of test samples revealed higher content of phytochemical constituents. The acetone extract of *P. aurata* showed strong DPPH radical scavenging activity with IC<sub>50</sub> value of 93.339 µg/mL. Similarly, the methanol extract of *U. bismolliuscula* and *R. taitensis* showed maximum ABTS scavenging potency and phosphomolybdenum reduction activity and with the highest values of 85.6 ± 0.5 TE/g and 150.68 ± 1.15 mg AA/g respectively. LCMS analysis of lichen extracts exhibited the secondary metabolites containing atranorin, sekikaic acid and usnic acid. It is concluded that these lichen species might be a valuable nutraceutical agent. Thus, the lichen extracts may further be assessed to toxic assays on experimental animals to confirm their biological activities.

**Keywords:** Lichens, Nutritional, Phytochemicals, Antioxidants, LC/MS

## 1. Introduction

Lichens contain mycobiont that has a defensive role and photobiont that supplies food. They have historically been used as spices in food, dyes in textile, odorant in perfume industry (Kosanic *et al.*, 2014). In our human body, singlet oxygen, hydroxyl (OH<sup>•</sup>) radicals, super oxide (O<sup>2-</sup>) anions and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) are biological Reactive Oxygen species (ROS) that induce oxidative stress and trigger degenerative diseases such as cancer, premature aging, diabetes, metabolism disorders and heart disorders (Kosanić & Ranković 2010). The lethal effect of radiation and pollutants and other exogenous factors are attributed to the formation of free radicals in biological systems. Lichen secondary metabolites such as depsides dibenzofuran-heterocyclic aromatic compound, depsidones-ester and pulvinic acid are all natural phenolic compounds were described to have antioxidant activities, and hence attracted considerable attention (Nguyen *et al.*, 2019).

Proximate elements of carbohydrates, proteins, fats and antioxidant properties of secondary metabolites of phenols are naturally found in lichens. Synthetic agents mostly

used as conventional chemical antioxidants have been pronounced to pose side effects such as inflammation and cell damage. Therefore, there has been the global interest in finding promising alternative antioxidant compound from natural sources. Conventionally, following solvents with different polarities such as petroleum ether, chloroform, acetone and methanol have commonly been employed to extract compounds from lichen thallus (Biney *et al.*, 2020) which deliver the dissolved compounds to diffuse into the cell through membrane pores and help to trigger its action. They have the property of low viscosity and high diffusivity (Biney *et al.*, 2020).

Although intensive and extensive assessment on the nutraceutical properties of most plants have been documented, little is known about the lichens in all over the World. Two such lichens include *Parmotrema tinctorum* and *Pseudocyphellaria aurata*, which are temperate foliose lichen. The relation of *Parmotrema tinctorum* with other spices in cooking is interesting and the use of this species may often add flavour in many dishes such as vegetables and meat by indigenous groups in Nepal and India (Upreti *et al.*, 2005). *Pseudocyphellaria aurata*, also known as specklebelly, belonged to the family Lobariaceae, while members of genus *Parmotrema* called

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as “ruffle lichens” belonged to Parmeliaceae. The tea prepared from *Pseudocyphellaria aurata* is considered to be beneficial for the health, particularly for the intestinal tract in curing indigestion. This has been used in Madagascar. The *Ramalina taitensis*, used to synthesize chemical compounds such as tannins, alkaloids, and saponins that are for biological functions, including antioxidant role and defence against microbial agents (Chowdhery 2014). The lichen substances of *Usnea bismolliuscula* play a great role in inhibition of tyrosine activity in medicine (Rajeswari *et al.*, 2019). Two of these species include which are fruticose lichens in the family Ramalinaceae and Parmeliaceae respectively. The members of the genus *Usnea* are called old man’s beard lichens and *Ramalina* are called strap lichens. These lichens have shown many health beneficial properties, such as possessing antibiotic, antioxidant, antiviral, antitumour, anti-allergenic and anti-inflammatory properties (Shrestha & St. Clair 2013). This has aroused further interest in the study of these lichens. Therefore, in the present study, the proximate chemicals, antioxidant property, and LC/MS analysis of methanol and acetone–water extracts of *Parmotrema tinctorum*, *Pseudocyphellaria aurata*, *Ramalina taitensis*, *Usnea bismolliuscula* were investigated to document the potential biological activities of these lichens. The results indicated that these lichen possess very high antioxidant potentiality, which gave immense scope for further exploration in nutritional and phytochemical studies.

## 2. Materials and methods

### 2.1. Collection and identification of lichen samples

Fresh thalli of *Parmotrema tinctorum*, *Pseudocyphellaria aurata*, *Ramalina taitensis*,

*Usnea bismolliuscula* were collected from Kodaikanal hills of Western Ghats, Southern India, during the winter season (January 2019). The elevation of hills was measured about 2130 meters. Identification of lichen was carried out at Biomedical Research lab, Bharathiar University, Coimbatore, Tamil Nadu, India. Voucher specimens were deposited at lichen herbarium centres, Bharathiar University, Coimbatore, India and CSIR-National botanical research Institute NBRI, Lucknow, India. The lichen specimens were identified by matching with the identification keys documented in Awasthi’s identification key manual (Awasthi 2007). The species were chemically identified with the help of conventional spot tests (K, C, I, KC) (Orange *et al.*, 2001). Lichen thalli were rinsed under flowing tap water to wash away soil particles. The samples were air-dried and left under sun shade to remove moisture. The air dried samples were blended into powder using a mixer and used for further analysis.

### 2.2. Nutritional studies

#### 2.2.1. Proximate chemical composition

Standard procedure was followed for the proximate chemical composition analysis of phytochemicals such as proteins, carbohydrates, total phenolics, amino acids, and total flavonoids (Lowry & Rosebrough 1951) lipids and fibre contents (AOAC 1990).

#### 2.2.2. Amino acids analysis

Amino acid content in the lichen extract was quantified according to the procedure described by Mohapatra *et al.* (2019). Briefly, a known amount of 100 mg of blended sample was acid treated with 10 mL 6 N HCl. The reaction mixture was digested in an oven at 120°C for 24 h. The aliquots were filtered out using the filter paper Whatman No. 1. The content was subjected to a vacuum flash evaporator. The amino acid concentrate was acid treated using 0.05 N Hydrochloric acid and filtered again using a 0.45 µm membrane filter and a volume of 20 µL was passed by injection into an amino acid analyzer (Shimadzu LC-10AS HPLC, McKinley Scientific, Sparta, NJ, USA) fitted with an ion exchange resin (styrene divinyl benzene co-polymer with sulphonic group) in column. The reference amino acid was injected to measure the amount of amino acid in the test sample. Amino acid standard was also run to calculate the concentration of amino acids in the samples.

#### 2.2.3. Mineral analysis by ICP-MS (Liu *et al.*, 2020)

During the last decades, ICP-MS (Inductively Coupled Plasma Mass Spectrometry) has emerged as the most promising technique for the analysis of trace elements of biological or environmental samples, and it is routinely used for the determination of up to eighteen elements in ionic research studies. The mineral analysis of lichen samples was done by the ICP – MS instrument of (Nex Ion 300 X, Perkin Elmer, USA). The powdered lichen material (200mg) was digested at 80°C with 10 mL of tri-acid (Nitric acid, Sulphuric acid and per chloric acid, 9:2:1) This test was performed by following the method of Lie *et al.* (2020). After digestion, all the samples were made up to 100 mL and aspirated into ICP – MS (Nex Ion 300 X, Perkin Elmer, USA).

### 2.3. Preparation of solvent extract

The extraction of lichen sample was done using the Soxhlet apparatus. The powdered lichen samples were loaded in pockets of thimbles and introduced in the extractor. Then, extraction was done separately using 250 mL solvents in an increasing polarity such as petroleum ether, acetone, methanol and water and heated up to 80°C for 8 h. Then, the extracts were concentrated using rotary vacuum evaporator (Superfit make, India). The concentrate thus obtained was weighed and stored in refrigeration at 4°C. Three samples from each solvent extract were examined for phytochemical and biological activities to assess variations in the test analysis. The presence of different lichen substances in each extract justifies the use of different solvents in an increasing polarity for extraction.

### 2.4. Quantitative analysis of phytochemicals.

#### 2.4.1. Estimation of total phenolic content (TPC)

The procedure is based on the method given by Ng and See (2019); Gaafar *et al.* (2019). Each sample was tested by adding 100 µL of extract (1 mg/mL) in 500 µL of 10% Folin–Ciocalteu reagent solution and incubating for 5 min at room temperature. This was followed by adding 2.5 mL of 7.5% sodium carbonate solution and incubating in dark for 45 min. The colour changes in test tubes were measured in spectrophotometer at 760 nm. The concentration ranges between 0–100 µg/mL of the

suggested Gallic acid standard were analysed to obtain the calibration curve. The mean  $\pm$  standard deviation of readings was calculated, and the results of TPC were represented in milligrams of gallic acid equivalents ( $\pm$ sd) per g of the lichen extract.

#### 2.4.2. Estimation of total flavonoid content (TFC)

The TFC procedure is based on Abdel-Mawgoud et al. (2019); Ng and See (2019) technique. About 500  $\mu$ L of the extracts were mixed with 300  $\mu$ L of 5% sodium nitrate solution followed by 300 $\mu$ L of 10% Aluminium chloride and incubated at room temperature. The content of the tube was added with 4% sodium hydroxide. As standard, a test tube containing Rutin was used to obtain calibration curve. From each extracts, triplicate samples were examined separately to find out any variations in the assessment of TFC and the mean value was calculated. The results were represented as Rutin Equivalents ( $\pm$ sd).

### 2.5. In vitro antioxidant assays

#### 2.5.1. DPPH radical scavenging activity

The 1,1-diphenyl-2-picrylhydrazine (DPPH) radical scavenging activity was determined bestowing to the procedure of Ng and Rosman (2019); Sedjati et al. (2020). Stable 1,1-diphenyl-2-picrylhydrazyl (DPPH) was assessed to examine the free radical capturing ability of H<sup>+</sup> ions. DPPH has the potency to donate H<sup>+</sup> ions to atoms that can give rise to free radicals of hydrogen. Reduction in the amount of DPPH molecule determines the free radical scavenging ability. It was analysed by detecting the absorbance readings at 570 nm. Reading in the absorbance decreases as the free radicals scavenged by the active substances of the lichen extracts increases. The percentage of inhibition for lichen extract was compared with the commercial ascorbic acid standard. The reaction mixture contained 100  $\mu$ L of aliquots with different concentrations of extracts. The positive control has different concentrations of ascorbic acid standard added with 3.0 mL of DPPH solution. The mixture was left in dark for 30 min at room temperature. The absorbance of the sample was read at 517 nm against a blank (methanol). The average of three readings was calculated. The following formula determines the results of half the inhibition concentration value (IC<sub>50</sub>) for DPPH radical scavenging ability of test samples:

$$X = [(A_0 - A_1) / A_0] \times 100$$

Where X= % Inhibition, A<sub>0</sub> is the absorbance of the control and A<sub>1</sub> is the absorbance of the sample extracts/standard.

#### 2.5.2. Superoxide anion radical scavenging activity

The principle of this assay is based on the ability of lichen extracts to inhibit the formation of formazan by scavenging the superoxide radicals present in riboflavin–light–NBT system according to the procedure described by (Thangaraj, 2016). About 3 mL reaction mixture contained 50mM Sodium phosphate buffer pH (7.6), riboflavin 20 $\mu$ g, EDTA (12mM), NBT (0.1 mg) and 100 $\mu$ L sample of lichen extract. The absorbance of the sample was measured at 590 nm against a blank (methanol). The mean value of triplicate readings was recorded. The percentage of half the inhibition concentration value (IC<sub>50</sub>) of superoxide anion generation was calculated by following equation:

$$X = [(A_0 - A_1) / A_0] \times 100$$

Where X= Superoxide anion scavenging activity, A<sub>0</sub> is the absorbance of the control and A<sub>1</sub> is the absorbance of the sample extracts/standard

#### 2.5.3. ABTS<sup>+</sup> (2, 2'-Azino-Bis-3-Ethylbenzothiazoline-6-Sulfonic Acid) radical scavenging activity

The principle of ABTS scavenging assay is based on the ability of lichen extract to reduce ABTS and lead to the decolouration of a green color complex to colourless. About 1 mL of diluted ABTS solution was transferred to the tube containing an aliquot of lichen extract (Rutin as standard) followed by adding 2.4 mM of potassium per sulphate and left in the dark. Triplicate determinations were calculated and the results of ABTS scavenging activity were read at 734 nm against a blank ethanol. The result was represented in mM of Trolox equivalents per g of the lichen extract (Makawy et al., 2019).

#### 2.5.4. Phosphomolybdenum reduction assay

The antioxidant potency of the lichen extracts was assessed by the phosphomolybdenum reduction assay (Zengin et al., 2015). The assay was based on the ability of a substance in lichen extract to reduce phosphomolybdic acid to blue complex phosphomolybdenum. A reagent solution was prepared containing ammonium molybdate (4 mM/L) and sodium phosphate (0.6 mM/L). Exactly, 250  $\mu$ l of the extracts was added with 3 mL of the reagent solution. The tubes with reaction mixture were incubated at 95°C for 90 min. The absorbance of the reaction mixture was measured at 695 nm using spectrophotometer against blank (0.3 mL methanol and 3 mL reagent). The ascorbic acid was used as standard (1mmol/L ascorbic acid in DMSO). The assay was done in triplicates.

### 2.6. Liquid Chromatography coupled to Mass Spectrometry (LC/MS) analysis

For studying LC/MS method followed by Zhang et al. (2019), the lichen substances obtained from the methanol extract of lichen samples were identified using Linear Trap Mass Spectrometer (Thermo Scientific). The extracts containing 5  $\mu$ l were injected into a UPLC(R)BEH C18 1.7 $\mu$ m-(2.1x100mm)) on a Water Xevo TQD 2000 C18 column which consists of 3000 nano system. The flow rate was followed at an interval of time 0.3 ml/min with the gradient elution program. The solvent mixture containing 70%-Acetonitrile : 30% Formic acid in water served as mobile phase and total run over time was automated to flow in 10.0 min. The compounds were identified using peaks and collected by scanning at wide wavelengths ranges between 100 and 1000 atomic mass units over the mass/charge number (m/z) (Xcalibur version 4.0). The LC/MS profile of the prevailing lichen compounds were compared with the standard Chemindex lichen data base. The mass spectral data for each compound was standardized to ensure the most favourable ion transfer conditions, ionization and obtained optimum peak of both the precursor and fragment ions. The source peak was compared for its identity for all the lichen samples.

### 2.7. Statistical analysis

All tests were prepared in triplicates. Data are documented as mean  $\pm$  standard error. The result was analysed statistically. The SPSS version 20.0 was used for this purpose. The one way ANOVA approach by Duncan's

test was followed. Mean values at  $p < 0.05$  were found statistically significant. For the graphical representation, R software version 3.5.1 was employed.

### 3. Results

#### 3.1. Proximate composition of samples.

##### 3.1.1. Moisture and ash contents of samples.

The moisture and ash contents of *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolliuscula* were investigated and results are presented in Table 1. The availability of moisture content was found to be significant in lichen

**Table 1.** Proximate chemical composition of Lichen samples

Parameters	Total carbohydrates (mg glucose equivalents/g sample)	Total proteins (mg BSA equivalents/g sample)	Total starch (mg glucose equivalents/g sample)	Energy (%)	Moisture content (%)	Ash content (%)
<i>P. tinctorum</i>	64.6 ± 0.01	25.78 ± 0.03	0.17 ± 0.03	82.85	10.5	13.93
<i>P. aurata</i>	31.63 ± 0.3	11.65 ± 0.11	0.14 ± 0.09	54.75	9.3	1.93
<i>R. taiensis</i>	50.12 ± 0.1	14.10 ± 0.5	0.12 ± 0.06	74.89	10	4.41
<i>U. bismolliuscula</i>	58.52 ± 0.02	24.95 ± 0.03	0.13 ± 0.05	60.05	9.8	2.43

##### 3.1.2. Determination of nutrient contents

The proximal composition of the lichen extracts such as *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolliuscula* are shown in Table 1. The total carbohydrates content was found highest in *P. tinctorum* (64.6 ± 0.01 mg GE/g sample), whereas *P. aurata* (31.63 ± 0.3 mg GE/g sample) registered lower carbohydrate content. The *P. tinctorum* had a maximal amount of total protein content (25.78±0.03mg BSAE/g sample) when compared to other lichens. Moreover, the lichen species, *R. taiensis* exhibited highest starch content. The *P. tinctorum* yielded a high energy level (82.85%) in the analyzed lichen extracts. The differences between the activity of standard used in each antioxidant assay was compared with previous published papers. Comparison of standard values with previous paper's standard values for the same assay validated moisture content results in this study. The present study has shown that the protein content of lichens is considerably low (foliose 1.87% and fruticose 1.9%) compared to the protein content of *P. pseudotinctorum* (16.2%) (Vinayaka *et al.*, 2010).

##### 3.1.3. Mineral composition of lichen samples

Mineral composition of the *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolliuscula* were analysed and presented in Table 2. Among all minerals tested, calcium

extract at any stage of nutritive analysis. The lichen moisture content was carried out in triplicates. It was found that the *P. aurata* showed lichen with lowest moisture content (9.3%) as compared to other lichen extracts. The *U. bismolliuscula* showed moderate moisture content (9.8 %). The result findings of *P. tinctorum* showed maximum moisture content of 10.5 % followed by *R. taiensis* with 10%. Similarly, the ash content was calculated and the results indicated positive effect with the lichens. *Parmotrema tinctorum* extracts were compared with the values of other lichen samples which showed highest level of total ash content which was up to 13.93%.

was found to be the highest in all lichen extracts followed by potassium. The third highest element was magnesium in *R. taiensis* (710 ppm) and *U. bismolliuscula* (530 ppm), whereas the extract of *P. tinctorum* (710 ppm) and *P. aurata* (660 ppm) showed aluminium as the third highest element. Therefore, the mineral content of *R. taiensis* and *U. bismolliuscula* was found to be in the order of calcium > potassium > magnesium and aluminium but the extract of *P. tinctorum* was in the order of calcium > potassium > magnesium and *P. aurata* was in the order of aluminium > calcium > potassium > phosphorous. The extract of *R. taiensis* was found to possess more amount of calcium content (3800 ppm). The extract of *P. tinctorum* was found to have the highest potassium content. The aluminium content was high in the extract of *P. tinctorum* with the value of 710 ppm. In the case of *P. aurata*, the iron and magnesium and zinc contents showed higher concentrations than other lichen sample extracts. The mineral content of cobalt and cadmium were the least quantity of all lichen extracts.

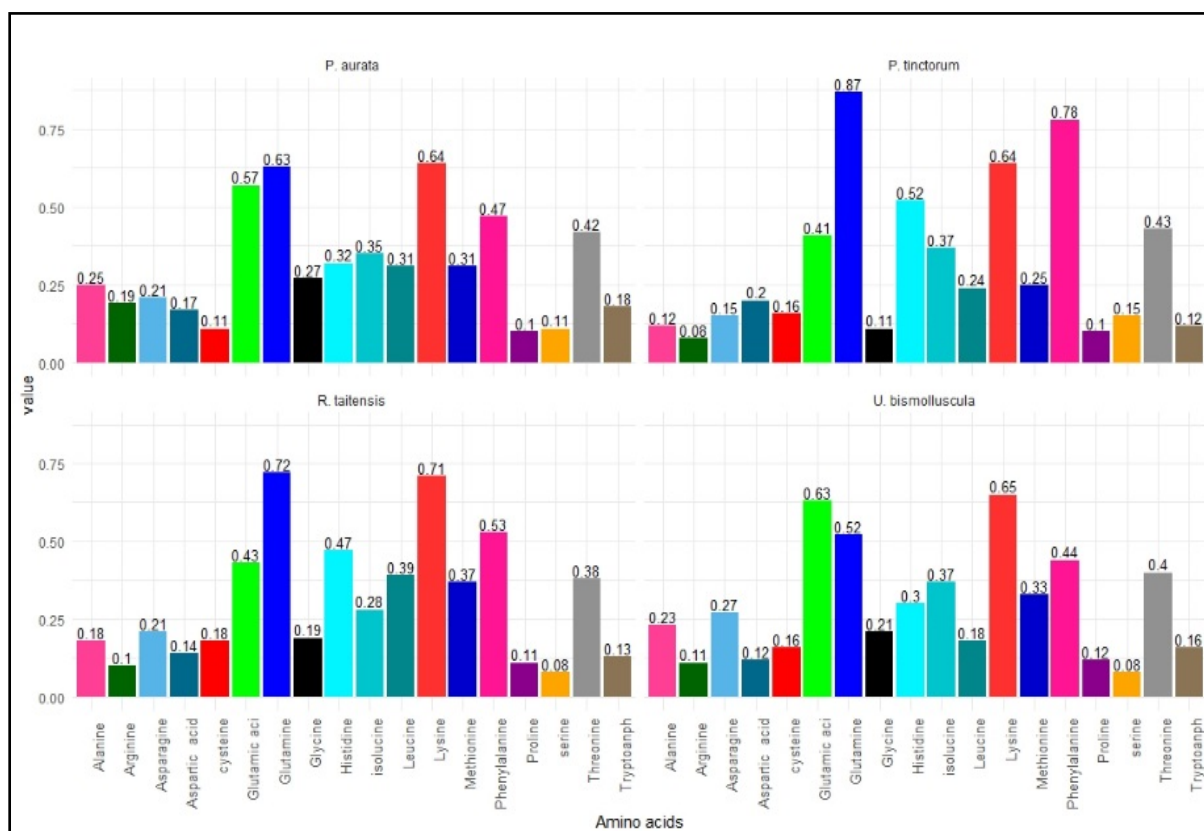
Vital minerals such as calcium (3800 ppm) and magnesium (710 ppm) were abundant in *R. taiensis*, and Sodium (200 ppm) and phosphorous were maximum in *P. tinctorum* compared to other lichens. Trace elements like Sr, Fe, B, Zn, Al, Si, Mn, and Cu also detected in all the analyzed lichens.

**Table 2.** Quantitative mineral element content of phytochemicals in lichen extracts

	<i>P. tinctorum</i> (ppm)	<i>P. aurata</i> (ppm)	<i>R. taitensis</i> (ppm)	<i>U. bismolluscula</i> (ppm)
Ca	1500	2500	3800	2000
Mg	510	420	710	530
Na	200	180	140	80
Sr	19	15	25	10
K	2000	1500	1600	1000
Fe	239	390	290	150
B	45	30	23	37
Zn	15	25	13	10
Al	710	660	340	460
Si	230	260	320	290
P	460	430	180	110
Mn	90	82	74	56
Cu	3.5	4.6	1.5	1
Co	0.1	0.4	0.5	0.2
Cr	0.2	0.7	1.2	1
Ni	1	1.2	1.4	1.2
Pb	40	28	30	17
Cd	0.1	0.3	0.5	0.2

### 3.1.4. Amino acids profiling of lichen samples

The amino acid composition of the lichen sample is shown in Figure 1. It was found that the proteins of the lichen contained adequate levels of amino acids. A total number of eighteen amino acids were identified from the studied samples. The estimated value of eight essential amino acids were found to be detected in all the four lichens. The amino acid glycine (0.27mg/g) was a dominant in *P. aurata* followed by alanine (0.25mg/g), arginine (0.19mg/g), tryptophan (0.18mg/g), aspartic acid (0.17mg/g). Glutamine (0.87mg/g), histidine (0.52mg/g), isoleucine 0.37mg/g, serine (0.15mg/g) and threonine (0.43mg/g) were higher in *P. tinctorum*. *R. taitensis* had its dominant amino acids of cystine (0.18mg/g), isoleucine (0.37mg/g), leucine (0.39mg/g), lysine (0.71mg/g), methionine (0.37mg/g) and phenylalanine (0.53mg/g). The amino acids of asparagine (0.27mg/g), glutamic acid (90.63mg/g) were more abundant in *U. bismolluscula*.

**Figure 1.** Amino acid profile of lichen extracts

### 3.1.5. Total phenolic content of lichen samples

Phenol plays an important role in scavenging free radicals present in the body. The TPC in *P. tinctorum*, *P. aurata*, *R. taitensis* and *U. bismolluscula* extracts were compared with the linear gallic acid standard curve. The phenolic amounts found in test samples of different lichen extracts were statistically evaluated and findings are shown in Table 3. The methanol extract of all the four lichen species showed a high amount of phenolic compound. In *R. taitensis*, it was  $86.14 \pm 1.42$  mg GAE/g extract while *U. bismolluscula* had  $55.60 \pm 1.11$ mg

GAE/g followed by *P. aurata* with the value of  $40.4 \pm 1.07$  GAE/g extract and *P. tinctorum* had  $39.5 \pm 1.29$  mg GAE/g extract. The aqueous extracts of *U. bismolluscula* showed the lowest phenolic content with the concentrations of  $4.9 \pm 1.31$  GAE/g extract. The study was carried out following the method of Ng et al. (2020b) and Lapido et al. (2011). The considerable variation in total phenolic content was observed in our present study with the highest concentration of  $86.14 \pm 1.42$  mg GAE / g extract in the methanol extract of *R. taitensis*.

**Table 3.** Estimation of total phenolics, flavonoids and ABTS radical scavenging activities, phosphomolybdenum, of lichen samples

Samples	Extracts	Total phenolics mg GAE / g extract	Total flavonoids mg RE / g extract	ABTS (mM TE/g extract)	Phosphomolybdenum (mg AAE/ g extract)
<i>P. tinctorum</i>	Petroleum ether	7.38 ± 0.10 <sup>j</sup>	8.63 ± 0.06 <sup>l</sup>	29.11 ± 1.06 <sup>n</sup>	27.78 ± 1.18 <sup>i</sup>
	Acetone	39.5 ± 1.29 <sup>c</sup>	17.6 ± 1.08 <sup>e</sup>	50.2 ± 4.9 <sup>f</sup>	37.00 ± 0.53 <sup>h</sup>
	Methanol	26.4 ± 1.15 <sup>d</sup>	25.8 ± 1.10 <sup>d</sup>	73.7 ± 5.2 <sup>b</sup>	42.56 ± 1.35 <sup>g</sup>
	Water	9.15 ± 1.21 <sup>l</sup>	5.9 ± 1.07 <sup>n</sup>	35.62 ± 0.03 <sup>i</sup>	31.62 ± 0.29 <sup>j</sup>
<i>P. aurata</i>	Petroleum ether	11.26 ± 0.03 <sup>h</sup>	7.73 ± 0.05 <sup>m</sup>	17.6 ± 6.10 <sup>o</sup>	27.35 ± 1.74 <sup>i</sup>
	Acetone	16.3 ± 2.04 <sup>f</sup>	10.6 ± 1.07 <sup>c</sup>	42.7 ± 5.11 <sup>g</sup>	50.68 ± 1.15 <sup>f</sup>
	Methanol	40.4 ± 1.07 <sup>b</sup>	26.9 ± 1.14 <sup>j</sup>	62.3 ± 15.1 <sup>c</sup>	68.54 ± 1.64 <sup>d</sup>
	Water	13.8 ± 0.01 <sup>g</sup>	3.6 ± 1.03 <sup>o</sup>	37.15 ± 0.04 <sup>h</sup>	59.74 ± 2.03 <sup>e</sup>
<i>R. taiensis</i>	Petroleum ether	12.01 ± 1.02 <sup>h</sup>	14.4 ± 1.088 <sup>g</sup>	29.5 ± 5.06 <sup>m</sup>	55.81 ± 0.08 <sup>f</sup>
	Acetone	22.87 ± 0.12 <sup>e</sup>	11.7 ± 1.10 <sup>f</sup>	30.4 ± 1.9 <sup>l</sup>	87.26 ± 0.39 <sup>b</sup>
	Methanol	86.14 ± 1.42 <sup>a</sup>	17.04 ± 0.04 <sup>h</sup>	53.12 ± 6.5 <sup>e</sup>	150.68 ± 1.15 <sup>a</sup>
	Water	6.3 ± 1.06 <sup>k</sup>	9.5 ± 1.08 <sup>k</sup>	14.2 ± 7.4 <sup>p</sup>	63.24 ± 0.391 <sup>e</sup>
<i>U. bismolluscula</i>	Petroleum ether	12.58 ± 0.082 <sup>h</sup>	11.36 ± 0.05 <sup>i</sup>	35.1 ± 6.1 <sup>j</sup>	72.47 ± 2.15 <sup>c</sup>
	Acetone	14.7 ± 1.04 <sup>e</sup>	30.05 ± 1.17 <sup>b</sup>	60.3 ± 14.2 <sup>d</sup>	51.196 ± 0.78 <sup>f</sup>
	Methanol	55.60 ± 1.11 <sup>a</sup>	31.2 ± 1.10 <sup>a</sup>	85.6 ± 0.5 <sup>a</sup>	87.78 ± 0.48 <sup>b</sup>
	Water	4.9 ± 1.31 <sup>k</sup>	10.4 ± 1.01 <sup>j</sup>	32.3 ± 2.19 <sup>k</sup>	71.93 ± 0.25 <sup>c</sup>

Values are mean of triplicate determination (n=3) ± standard deviation, % - % of Inhibition.

GAE-Gallic Acid Equivalents, RE- Rutin, TE- Trolox Equivalents. AAE- Ascorbic Acid Equivalents.

Statistically significant at  $p < 0.05$  where  $a > b > c > d > e > f > g > h > i > k > l > m > n > o > p$ , in each column.

The results of the present study are strongly supportive in agreement to the findings of Rice-evans *et al.* (1995) and Singleton *et al.* (1999) that the TPC content of lichen extracts are responsible for antioxidant and biological activities of lichen extracts. Ganesan *et al.* (2015). Observed that the TPC value was significantly higher in the extract of benzene (154.2 mg GAE/g) than in other *P. tinctorum* extracts and reported that the availability of phenolic content sources could trigger the activity of lichen metabolites which probably play a subtle role in antimicrobial and antioxidant activities. The results are not in agreement with the report given by Ganesan *et al.*, (2015). The relative TPC values of lichen or tested plants may be varied due to age of the plant, methods of extraction procedure, choice of solvents, number of replicates used for investigation etc. Singleton *et al.* (1999) observed the number of phenolic groups in their native structure and reported that this may have the major reason for difference found between biological activities of lichen extracts. Rice-Evans *et al.* (1995) reasoned phenolic compound has the high potency of antioxidants because it acts as the single oxygen quenchers and oxygen donors. As reported, the acetone extracts of *P. austrosinense* and *P. tinctorum* exhibited high TPC corresponding to its high free radical scavenging activities (Kalidoss *et al.*, 2019).

### 3.1.6. Total flavonoid content of lichen samples

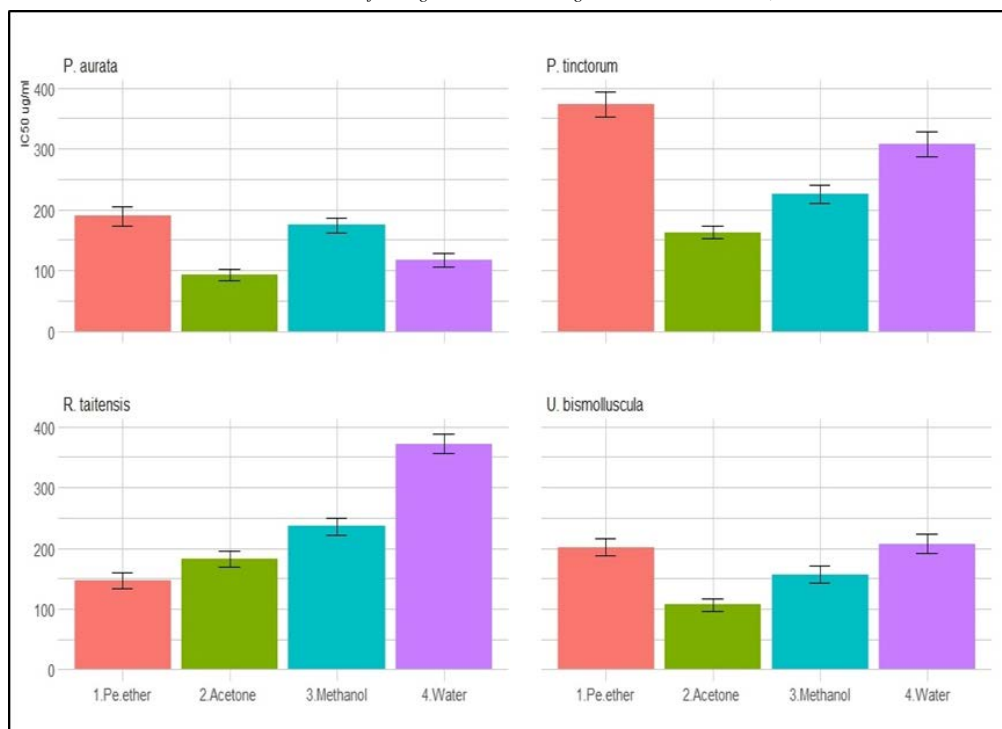
The TFC was determined in *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolluscula* and the results are presented in the Table 3. The highest TFC was observed in methanol extracts of *U. bismolluscula* (31.2 ± 1.10 mg RE/g extract) followed by *P. aurata* (26.9 ± 1.14 mg RE/g extract) and *P. tinctorum* (25.8 ± 1.10 mg RE/g extract) and *R. taiensis* (17.04 ± 0.04 mg RE/g extract). The aqueous extracts of *P. aurata* was found to be lower (3.6 ± 1.03 mg RE/g

extract). The results reported by Ng, Koick and Yong (2020a) are strongly supportive in agreement to the present investigation that the flavonoid and phenolic constituents of plant extracts are considered to be beneficial for health particularly for the antioxidant action. Thus, the use of lichen extracts for health benefits particularly for nutraceutical applications can be justified on the basis that these lichens were once thought to be medicinal is really due to their TPC, TFC and other chemical constituents.

### 3.2. In vitro antioxidant assays

#### 3.2.1. DPPH radical scavenging activity

The DPPH radical scavenging activities of different lichen extracts of *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolluscula* are shown in the Figure 2. The IC<sub>50</sub> values (concentration of sample required to scavenging 50% of free radicals) were calculated from the regression equation prepared from the concentration of the extracts versus percentage inhibition of free radical formation. A lower IC<sub>50</sub> value indicates the greater DPPH radical scavenging activity. Among the all the lichens analysed the acetone extract of *P. aurata* shows lower IC<sub>50</sub> value of 93.339 µg/mL than the acetone extract of *U. bismolluscula* (106.786 µg/mL), petroleum ether extract of *R. taiensis* 146.9266 µg/mL and acetone extract of *P. tinctorum* (162.488 µg/mL). The petroleum ether extract of *P. tinctorum* shows minimal antioxidant activity with the highest IC<sub>50</sub> value of 372.825 µg/mL. The IC<sub>50</sub> value for ascorbic acid (20.32 µg/mL) standard employed in DPPH assay of the present study is closer to the value given by Ng *et al.* (2020c). The present study showed significant half the inhibition concentration of ascorbic acid standard with the IC<sub>50</sub> value of 20.32 µg/mL as compared to 23.45 mmol/L showed by Ng *et al.* (2019).

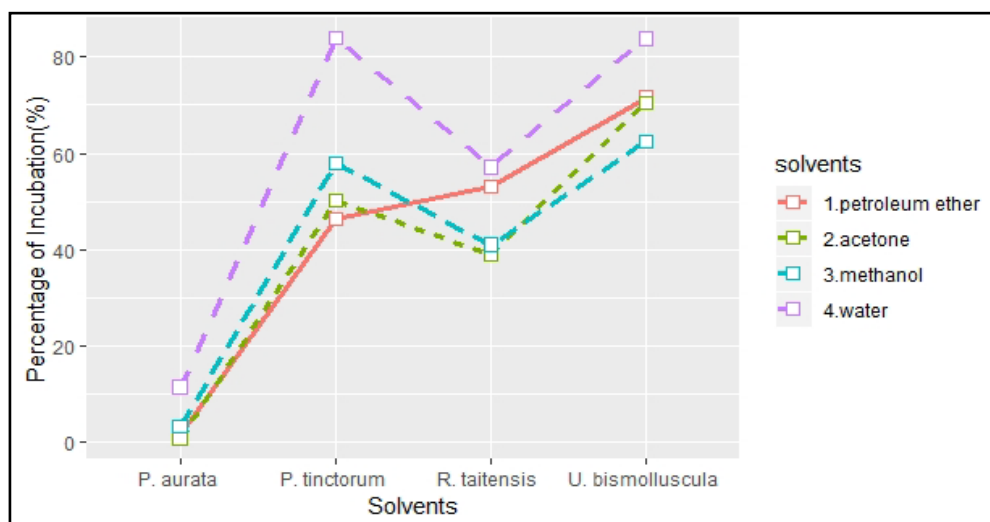


**Figure 2.** DPPH radical scavenging activity of lichen extracts

### 3.2.2. Superoxide anion radical scavenging activity

Superoxide anion radical scavenging activities of *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolluscula* were analysed, and the results are shown in Figure 3. The extracts were found to be containing high scavenging potential of superoxide radicals generated in riboflavin-NBT- light system. The water extract of *P. tinctorum*, was

showing higher (83 %) scavenging potential than the other extracts. The minimum activity with the IC<sub>50</sub> value was found in acetone extract of *P. aurata* (0.78%). The result proves that the percentage of inhibition is based on the concentration of the bioactive phenolic compounds present in solvent extract.



**Figure 3.** Superoxide radical scavenging activity of lichen extracts

### 3.2.3. ABTS<sup>+</sup> radical scavenging activity

The extracts of *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolluscula* showed their antioxidant activities with regard to ABTS scavenging abilities (Table 3). The methanol extracts of *U. bismolluscula* showed highest ABTS scavenging activity with the value of  $85.6 \pm 0.5$  mM TE/g extract. The methanol extract of *P. tinctorum* and *P. aurata* showed moderate activity with the values of  $73.7 \pm 5.2$  and  $62.3 \pm$

$15.3$  mM TE/g extract respectively. However, the methanol extracts of *R. taiensis* showed minimal scavenging values ( $53.12 \pm 6.5$  mM TE/g extract). The values of the various extracts of lichens for the ABTS scavenging activity falls in the order of *U. bismolluscula* > *P. tinctorum* > *P. aurata* > *R. taiensis* respectively. The standard rutin showed strong ABTS scavenging activity ( $111.3 \pm 0.52$  mM TE/g extract). Compared to that of standard rutin, moderate ABTS scavenging activity was observed in the methanol extract of *U. bismolluscula* (highest value of  $85.6 \pm 0.5$  mM TE/g extract) in the

present study. Similar study conducted by Ng *et al.* (2020a) showed 0.01 mg/mL for trolox standard as compared to  $111.3 \pm 0.52$  mM TE/g extract for rutin in the present study. Variations in the values are due to variations in the test standard.

#### 3.2.4. Phosphomolybdenum reduction assay

The phosphomolybdenum assay was mostly used to measure the total antioxidant activity of substances and the results are presented in the Table 3. Among the different extracts *P. tinctorum*, *P. aurata*, *R. taiensis* and *U. bismolliuscula*, methanol extracts of *R. taiensis* showed highest value of reducing ability compared with other extracts ( $150.68 \pm 1.15$  mg AAE/ g extract). The methanol extracts of *P. tinctorum*, *P. aurata* and *U. bismolliuscula* were found to have moderate level of antioxidant activity with the concentrations of  $42.56 \pm 1.35$ ,  $68.54 \pm 1.64$  and  $87.78 \pm 0.48$  mg AAE/ g extract respectively. The rutin standard showed maximum activity ( $158.39 \pm 1.79$  mg AAE/ g extract) compared with lichen extracts in phosphomolybdenum reduction assay. The study carried out in methanol extracts of four lichen thalli showed maximum phosphomolybdenum activity when compared to the other solvent extracts. The phosphomolybdenum reduction value in methanol extract of *R. taiensis* is closer to the rutin standard.

## 4. Discussion

The nutritional constituents of two foliose and two fruticose forms of lichen species were documented as described by Hu *et al.* (2020); Rahman *et al.* (2018); Kekuda, *et al.* (2011); Joubert *et al.* (1982). The previous literature asserted that the nutritional contents of lichens such as tannins, flavonoids and phenolics are being used as medicine and can be useful to fortify them in human foods. Lichens and plants differ considerably in phytochemical constituents. Since many wild animals and mammals eat lichen as diets, it is vital to measure the nutritional content of lichens. The nutrient content of foliose lichen was comparable with fruticose lichen in all macro element parameters except ash content. The moisture content of the four different lichens was found to be in the order of *P. tinctorum* > *R. taiensis* > *U. bismolliuscula* > *P. aurata*. This may be the reason to suggest that the *P. tinctorum* weighs heavier than other lichens.

Compared with foliose forms, fruticose lichens were deficient in starch and ash content. Fruticose were also lower in energy than foliose lichen. Many lichen species are also shown to be eaten by many mammals such as red-backed voles, deer and squirrels and thereby may play an important role in economic botany as nutritional benefits to plants and animals.

Previous literatures have shown that organic solvents are generally used for extraction of lichen metabolites because lichen substances are either insoluble or partial soluble in water. Most organic solvent-soluble antioxidants are either bound to cortex or medulla in lichen thalli. Solvent extracts from test samples of four different lichen thalli responded effectively to various biological assays in the present investigation. The extract of a particular lichen substance influences the efficiency of biological assays and is based on the polarity of solvent.

According to Sarker *et al.* (2006), the use of organic solvent, methanol for extraction of metabolites to winged been seeds has been successful because these affected easy extraction process in many ways and that the methanol penetrated the thick seed coat structure and promoted fast recovery of plant nutrients. The mechanism behind the effect of solvent property may also depend on the electron-transfer kinetic mechanism and physiochemical properties (lipophilic and hydrophilic properties) of the phytochemical compounds. Previous records have shown that the ideal solvent examined for extracting the antioxidant compounds of *P. tetragonolobus* was methanol in antioxidant assays (Ng *et al.*, 2020b). This is in agreement with our findings that the methanol extract of all the four lichens had the highest radical scavenging activity and the values are comparable to the standard.

The use of petroleum ether and water had found negative effects in the present study on ABTS, DPPH, superoxide scavenging and phosphomolybdenum reduction assays by showing the weak activity (the highest IC<sub>50</sub> or low mM TE/mg values) with minimal antioxidants to scavenge radicals inhibiting cellular damage. The primary reason for lowest antioxidant activity may be attributed to the relative total antioxidant contents possessed in solvent extracts. Similarly, the considerable variation in types of substances or antioxidants has also been observed in the present investigation, thus they reacted differently to the range of solvents on the basis of their polarity. The radical scavenging activity suggested a contrast between the electron transfer kinetic mechanism of substances in solvent extracts. The DPPH radical scavenging power is attributed to the lipophilic antioxidants while, the ABTS antioxidant assay detects lipophilic and hydrophilic substances (Ng & See 2019) in the extracts indicating the differences in the antioxidant activity between these two scavenging assays.

The antioxidant activity may also be affected by enzymatic antioxidants. Previous literature indicated that the metabolite accumulation in plant part infers the induction of defence mechanism. Therefore, under abiotic stress, enzymatic antioxidants like, Superoxide dismutase and Quinone oxidoreductase are induced for scavenging reactive oxygen species (ROS). Superoxide dismutase play a significant role in the catalytic conversion of highly reactive superoxide anions to oxygen molecule while the quinone oxidoreductase enzymatic antioxidant involves in the conversion of highly reactive quinone to less reactive hydroxy quinones. This process of conversion of ROS is termed redox homeostasis defence mechanism (Ng *et al.*, 2014). The same reason may also be applicable to the present investigation that the level of enzymatic antioxidants in different solvent extracts might influence radical scavenging activities in antioxidant assays.

#### 4.1. Proximate composition of samples

All the macro and micro nutrients estimated were found to be higher in fruticose form than in foliose form of lichen (Table 1). The results showed that the former had the profuse development and growth when compared to the latter. The mean average total carbohydrate and protein contents were higher in fruticose (54.32 and 19.525 mg glucose equivalent respectively) than in foliose (48.115 and 18.715 % respectively) but there was no significant difference drawn between foliose and fruticose forms of



lichen in the amount of macronutrient. Kirkpatrick (1996) found that *Bryoria* spp., contained considerable quantities of soluble carbohydrates and pointed out that it had low fiber and phenolic contents. Robbins (Robbins 1987) found that the crude protein content of *Alectoria sarmentosa* lichen was just 2% and an outcome was nitrogen losses in feces of mule deer.

Starch content was found to be  $0.17 \pm 0.03$  mg glucose equivalents/g of samples in foliose followed by  $0.13 \pm 0.05$  mg in fruticose lichens. Similar to that of macroelements, the mean average ash content was more pronounced in foliose lichen (7.93%) followed by fruticose (3.42%) lichens. Moreover, such differences are determined experimentally, and the results were affected by environmental conditions such as air, pH of the substrata, temperature ...etc. There were no correlations between moisture and macro nutrient content in lichen growth forms (Table 1).

Thallus of *P. tinctorum* had a greater amount of ash content at 13.93% when compared to *R. taiensis* at 4.41%. The ash content of *U. bismolliuscula* thallus was found to be 2.43 % higher than 1.93% of *P. aurata*. The nutritional studies (Table 1) with thallus of foliose and fruticose lichen showed differences in starch, protein and total carbohydrate. This shows the significance of incorporating ash content in the analysis as majority of the nutritional content is included in the ash.

It follows from the results that foliose and fruticose lichen significantly differ in their nutritional values. Such differences are determined experimentally and may also be influenced by the nature of thallus.

#### 4.2. Mineral composition of lichen samples

Carbohydrates, proteins, fats and vitamin constituents are more common in all plants determining the nutritional quality of any forage. This macro nutrient together with micro elements form a fodder for reindeer, caribou, squirrels. These micronutrients have been described to have many physiological functions such as anti-oxidant defence activity, electrolytic activity, components of enzymes, gas transport, regulate cellular energy transduction and regulate physico-chemical processes. Therefore, these microelements play a vital role on all living matter. The result of the present study is in accordance with Vinayaka *et al.* (2013) reported the content of calcium was highest among all other elements in *U. pictoides*. Similar result was obtained by Kekuda *et al.* (2011) for the lichen extracts of *Everniastrum cirrhatum* where the content of calcium was highest among all micronutrients. The reason for the difference in microelements is probably due to variation in abiotic and biotic factors viz., habitats, age of lichen, solvent used for extraction procedure and origin of plant materials.

#### 4.3. Amino acids profiling of lichen samples

It was found that the lichen extracts contained sufficient amount of amino acids as compared to reports from WHO/ FAO 2007. The comprehensive outcome of this research culminates in the assumption that glutamine has a higher volume relative to all other amino acids in *P. tinctorum* in the present study. Supportive finding was recorded from Jäger and Weigel (1978) who reported that a conceivable intention for this similarity is that glutamine seems to play important role in nitrogen fixation and

*Pseudevernia furfuracea* (L.) Zopf was possessing high amounts of glutamic acid, aspartic acid, alanine, arginine, and taurine etc. Shelukheeva and Nikolaeva (2015) reported that amino acid content of *Cetraria islandica* and *C. laevigata* was found to possess 16 different amino acids, of which seven were essential. The present study detected 18 different amino acids from four different lichen samples.

#### 4.4. Estimation of total phenolic content (TPC)

Singleton *et al.* (1999) observed the number of phenolic groups in their native structure and reported that this may have the major reason for variation in biological activities of lichen extracts. Rice-Evans *et al.* (1995) reasoned phenolic compound has the high potency of antioxidants because it acts as the single oxygen quenchers and oxygen donors. As reported, the acetone extracts of *P. tinctorum* exhibited high phenolic content corresponding to its strong free radical scavenging activity.

#### 4.5. 4.1.5. Estimation of total flavonoid content (TFC)

In the present study, highest TFC was observed in the methanol extract of *U. bismolliuscula*  $31.2 \pm 1.10$  mg RE /g while the least was in the aqueous extract of *P. aurata*  $3.6 \pm 1.03$  mg RE /g. But, Ganesan *et al.* (2015) observed that the TFC of lichen extracts ranged from 20.16 to 50.72 mg RE/g, and the higher content was found in aqueous extract  $50.72 \pm 0.13$  and lower in ethanolic extract  $20.16 \pm 0.11$  mg RE/g. The results of the present study are in line with the Raj *et al.* (2014) reported that the total flavonoid and phenolic contents of *Parmotrema tinctorum* extracts were found to be  $5.82 \pm 0.96$  mg quercetin equivalents/g and  $9.67 \pm 0.73$  mg gallic acid equivalents/g of extract respectively.

#### 4.6. DPPH radical scavenging activity

DPPH scavenging assay was conducted to identify substantial antioxidant activity. DPPH is a free radical used to measure the scavenging potential of any phytochemical compounds. Among all four lichen extracts, the IC<sub>50</sub> value of acetone extract of *P. aurata* (93.339 µg/ml) is comparable to that of ascorbic acid (20.32 µg/ml). The DPPH scavenging results are supportive in agreement to Ristic *et al.* (2016) where lecanoric acid showed weak activity with the highest IC<sub>50</sub> value of 424.51 µg/mL. Thus, this suggested that the polarity of solvent extract that determines the DPPH radical scavenging activity of phytochemical compounds.

#### 4.7. Superoxide anion radical scavenging activity

The biological enzyme XO (xanthine oxidase) has significance in catalysing the bioconversion of hypoxanthine to xanthine and to uric acid while the SOD (superoxide dismutase) in superoxide into hydrogen peroxide and oxygen in biological tissues. The superoxide radical scavenging activity of the present analysis is in accordance with the Behera *et al.* (2016) who reported that the methanol extract of *U. ghattensis* exhibited strong activity with the value of 56%. The secondary metabolite of *U. ghattensis* contains usnic acid. The known standards BHA, BHT and quercetin showed scavenging activity with the value of 59, 68 and 47 % respectively. Kosanić *et al.* (2014) found that highest super oxide anion scavenging activity (67.37%) was observed in the acetone extracts of the lichen *Lasallia pustulata*. Aqueous extract of lichen

*Parmelia sulcata* showed the least scavenging activity (12.74%). The *P. tinctorum* aqueous extract of the present study showed the strong scavenging activity with the 87% while the weak activity was observed in acetone extracts of *P. aurata* with the lowest value of 0.78%.

#### 4.8. ABTS<sup>+</sup> radical scavenging activity

The comparison of the secondary metabolites of the tested lichen extracts and their free radical scavenging ability revealed a strong correlation with the previous study conducted by Raj *et al.* (2014). The solvent of choice greatly influenced the ABTS scavenging activity in extracts of all the four different lichen samples. This ethyl acetate extract of *P. tinctorum* exhibited significant antioxidant potential against ABTS with half the inhibitory concentrations of  $151.34 \pm 1.79$  mg/mL and the results are supportive in agreement to the previous literature depicted that the antidiabetic nutraceutical properties have been identified from *P. tinctorum* samples. In the present study, petroleum ether and aqueous extract of *P. tinctorum* showed less activity when compared with acetone and methanol extracts. The highest activity was shown in methanol extract ( $85.6 \pm 0.5$  mM TE/g extract) of *U. bismolliuscula* followed by the *P. tinctorum* extract ( $73.7 \pm 5.2$  mM TE/g extract). The results of ethyl acetate extract of *P. tinctorum* are supportive in agreement to the report given by Raj *et al.* (2014). Ganesan *et al.* (2017) found that methanol extract of *Ramalina inflata* (80 µg/ml) was significantly lower than benzene extract (100 µg/ml) of those recorded in the fructicose lichen. The results of the present study are consistent with the report of Ganesan *et al.* (2017), in that the antioxidant activity in the petroleum ether extract was lower than that of the other extracts. The results of ethanol extract of *U. longissima* are in accordance with the findings of Aydın *et al.* (2018) who reported that ABTS radical scavenging assay for *U. longissima* ethanol extract and ethyl acetate were found as  $73.31 \pm 0.007\%$  at 4000 µg/mL and  $54.92 \pm 0.010\%$  at 4000 µg/ml respectively. Compared with standard control, in the present study, ABTS radical scavenging activity of acetone extract of *U. bismolliuscula* was found to be high ( $60.3 \pm 14.2$  mM TE/g extract) while the petroleum ether and aqueous extracts exhibited weak activity. The disparity in ABTS scavenging activity of lichen extracts may be due to variation in environment, lichen species, assessment methods and types of solvents used by researchers (Kazazic *et al.*, 2016).

#### 4.9. Phosphomolybdenum reduction assay

In the present study, strong antioxidant activity was amounted to  $150.68 \pm 1.15$  mg AAE/ g for methanol extracts of *Ramalina taitensis*. Previous literature has

reported the antioxidant properties of solvent extracts of many species of lichen, but the present study is the first attempt to document the phosphomolybdenum reduction assay of four different lichen extracts of commonly available lichen species such as *P. tinctorum*, *P. aurata*, *R. taitensis* and *U. bismolliuscula*. The results showed that the methanol extract had a higher phenolic content (except *P. tinctorum*), which in turn exhibited strong antioxidant potency than the other extracts. The present result is in agreement with Dandapat and Paul (2019) who opined that methanol is an ideal solvent for extraction of lichen bioactive compounds for antioxidant activities. It is suggested that the increase in phenolic groups in chemical structure of lichen compounds can have corresponding increase in antioxidant activity. This is in conformity with the reports of Tomović *et al.* (2016).

The results of Manojlovic *et al.* (2012) are in accordance to the present study that the redox properties of phenolic content in methanol extract activated the neutralizing free radicals activity which play a subtle role in quenching triplet and singlet oxygen. They have shown that a significant difference in TPC was observed between chloroform ( $71.32 \pm 0.89$ ) and methanol extracts ( $79.2 \pm 0.59$  mg GA/g) of *Umblicaria cylindrica*, and it was correlated with their antioxidant activity of  $68.35 \pm 0.15$  µg AA/g and  $74.65 \pm 0.75$  µg AA/g respectively. Similarly, Tomović *et al.* (2016) reported the supportive results for total antioxidant capacity with the concentrations of  $91.52 \pm 0.34$  µg AA/g and  $71.5 \pm 0.29$  µg AA/g, for methanol and ethyl acetate extracts of *Cetraria aculeata*, respectively.

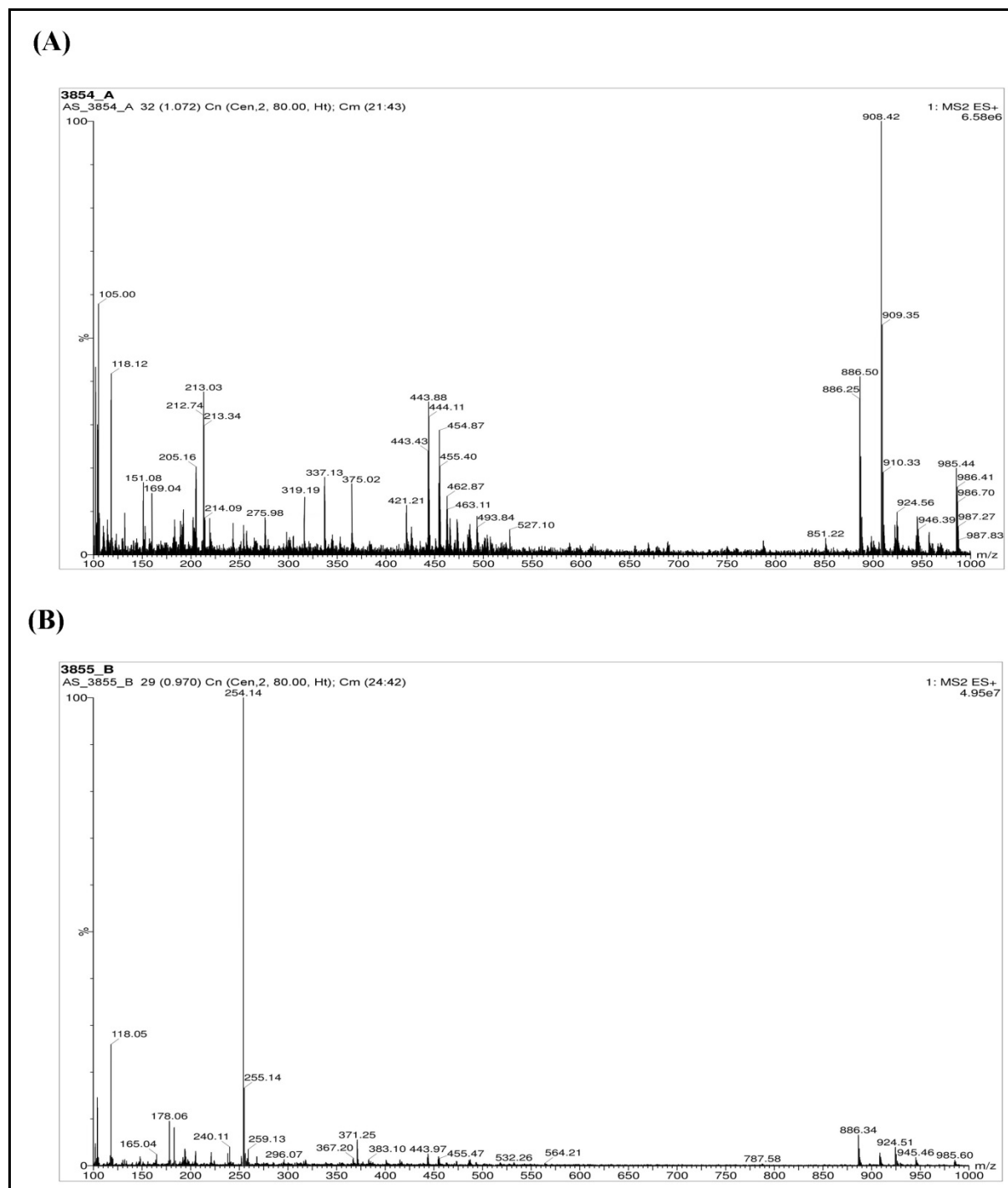
#### 4.10. Compound identification by Liquid

##### Chromatography Mass Spectrometry (LC/MS)

The results obtained from the studies indicated that methanol extract showed higher activity than other solvents. The Liquid Chromatography Mass Spectrometry study was conducted to find out the active lichen compounds in test samples and the chromatogram are shown in the Figure 4 & 5. Major peaks with relative Mass spectral data were observed from chromatogram. The lichen compounds were detected based on their (Chemindex) library data base. The major lichen substances like atranorin, sekikaic acid, usnic acid, and other compounds identified are presented in Table 4. The findings are in line with the previous literature (Musharraf *et al.*, 2015). Biologically important phytochemicals are present in *P. tinctorum*, *P. aurata*, *R. taitensis* and *U. bismolliuscula*. It can be suggested that the antioxidant activity is determined by their biologically active compounds in the solvent extract.

**Table 4.** Compound Identification of lichen extract by LCMS method

S.No	Compound Name	Exact mass (m/z)	Observed mass (m/z)	Precursor m/z	Molecular formula
<i>Parmotrema tinctorum</i>					
1.	Diethanolamine	105.14	105.00	117.01 -ve	C <sub>4</sub> H <sub>11</sub> NO <sub>2</sub>
2.	Succinic acid	118.09	118.12	119.03 +ve	C <sub>4</sub> H <sub>6</sub> O <sub>4</sub>
3.	D-Carvone	150.22	151.08		C <sub>10</sub> H <sub>14</sub> O
4.	Orsellinic acid	168.15	169.04	167.03 -ve 167.10 -ve	C <sub>8</sub> H <sub>8</sub> O <sub>4</sub>
5.	Eugenitol	206.19	205.16	205.04 -ve 207.06 +ve	C <sub>11</sub> H <sub>10</sub> O <sub>4</sub>
6.	Volemitol	212.2	213.03	211.08 -ve	C <sub>7</sub> H <sub>16</sub> O <sub>7</sub>
7.	Lecanoric acid	318.28	319.19	317.06 -ve 317.06 -ve	C <sub>16</sub> H <sub>14</sub> O <sub>7</sub>
8.	Methyl protolichesterinate	338.5	337.24		C <sub>20</sub> H <sub>34</sub> O <sub>4</sub>
9.	Atranorin	374.3	375.02	373.09 -ve	C <sub>19</sub> H <sub>18</sub> O <sub>8</sub>
10.	Thamnolic acid	420.3	421.21	419.06 -ve	C <sub>19</sub> H <sub>16</sub> O <sub>11</sub>
11.	Perlatolic acid	444.5	443.88	443.2 -ve	C <sub>25</sub> H <sub>32</sub> O <sub>7</sub>
12.	Lobaric acid	456.5	455.4	457.18 +ve	C <sub>25</sub> H <sub>28</sub> O <sub>8</sub>
13.	Leucotylin	460.7	462.87		C <sub>30</sub> H <sub>52</sub> O <sub>3</sub>
<i>Pseudocephalaria aurata</i>					
14.	Succinic acid	118.09	118.05	117.01 -ve 119.03 +ve	C <sub>4</sub> H <sub>6</sub> O <sub>4</sub>
15.	5,7Dihydroxy-6-methylphthalide	180.04	178.06		C <sub>9</sub> H <sub>8</sub> O <sub>4</sub>
16.	Chrysophanol	254.24	254.14	255.06 +ve	C <sub>15</sub> H <sub>10</sub> O <sub>4</sub>
17.	Conorobaridone	370.14	371.25		C <sub>21</sub> H <sub>22</sub> O <sub>6</sub>
<i>Ramalina taiensis</i>					
18.	Fumaric acid	116.07	118.05	115.00 -ve	C <sub>4</sub> H <sub>4</sub> O <sub>4</sub>
19.	D-Arabinose	150.13	151.04	149.04 -ve	C <sub>5</sub> H <sub>10</sub> O <sub>5</sub>
20.	Methyl orsellinate	182.17	183.06	181.05 -ve	C <sub>9</sub> H <sub>10</sub> O <sub>4</sub>
21.	Cholin sulphate	183.05	184.12		C <sub>5</sub> H <sub>13</sub> NO <sub>4</sub> S
22.	Methyl β-ornicolcarboxylate	196.2	197.08	195.06 -ve	C <sub>10</sub> H <sub>12</sub> O <sub>4</sub>
23.	Fukinanolide	234.33	236.14		C <sub>15</sub> H <sub>22</sub> O <sub>2</sub>
24.	Methyl porphyrilate	328.13	330.12		C <sub>17</sub> H <sub>12</sub> O <sub>7</sub>
25.	Usnic acid	344.3	345.08	343.08 -ve	C <sub>18</sub> H <sub>16</sub> O <sub>7</sub>
26.	3'-Dechlorolecideoidin	364.03	367.23		C <sub>17</sub> H <sub>13</sub> ClO <sub>7</sub>
27.	Sekikaic acid	418.4	419.35	417.14 -ve	C <sub>22</sub> H <sub>26</sub> O <sub>8</sub>
<i>Usnea bismolliuscula</i>					
28.	Diethanolamine	105.14	104.07		C <sub>4</sub> H <sub>11</sub> NO <sub>2</sub>
29.	Indolyl-3-acetic acid	175.18	175.04	176.06 +ve	C <sub>10</sub> H <sub>9</sub> NO <sub>2</sub>
30.	Olivetolmonomethylether	194.13	193.09		C <sub>12</sub> H <sub>18</sub> O <sub>2</sub>
31.	Ethyl everninate	210.23	211.11		C <sub>11</sub> H <sub>14</sub> O <sub>4</sub>
32.	Olivetolcarboxylic acid	224.25	225.17	223.09 -ve	C <sub>12</sub> H <sub>16</sub> O <sub>4</sub>
33.	Fukinanolide	234.33	236.14		C <sub>15</sub> H <sub>22</sub> O <sub>2</sub>
34.	Linoleic acid	280.4	279.27	279.23 -ve	C <sub>18</sub> H <sub>32</sub> O <sub>2</sub>
35.	Usnic acid	344.3	345.08	343.08 -ve	C <sub>18</sub> H <sub>16</sub> O <sub>7</sub>
36.	4-O-Methyl-5-dechlorovicanicin	362.09	363.10		C <sub>19</sub> H <sub>19</sub> ClO <sub>5</sub>



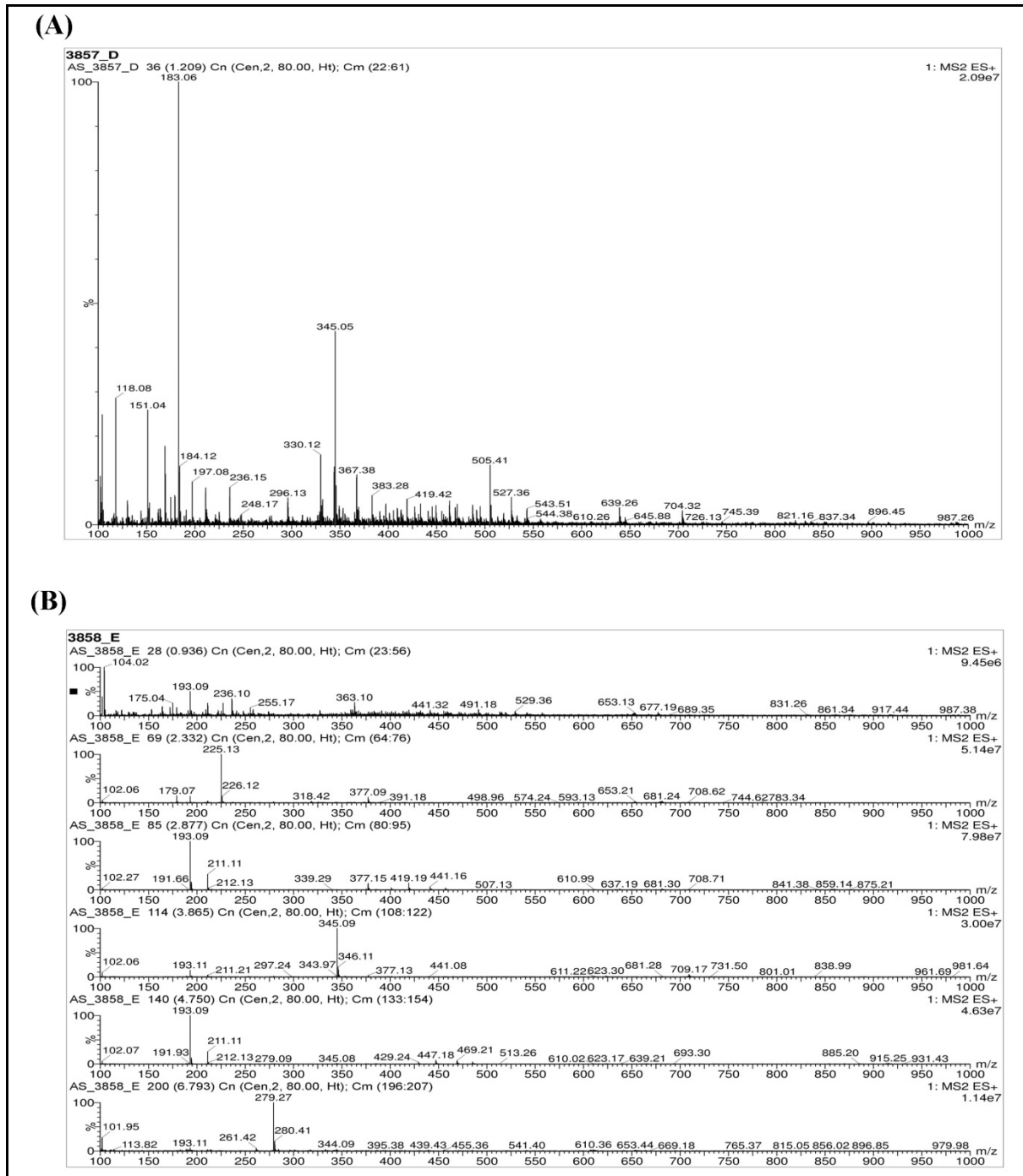


Figure 5. LC MS Chromatogram of A) *Ramalina taitensis* B) *Usnea bismolliuscula* extracts

## 5. Conclusion

The present research concluded that the extracts of *Parmotrema tinctorum*, *Pseudocyphellaria aurata*, *Ramalina taitensis*, *Usnea bismolliuscula* exhibited a variety of phytochemical constituents in solvent extracts. The moderate antioxidant activities of methanol extract depicted the possible application of these lichens in pharmacology. These can be a promising alternative to synthetic antioxidants. Additionally, it is suggested that the lichen substances such as atranorin, sekikaic acid, usnic acid detected in solvent extracts using LC/MS are responsible for radical scavenging activity. Further animal

studies are underway to assess the cytotoxicity effects of these lichen extracts.

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## Conflict of interest

The authors have no conflict of interest.

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